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Ground States of DCNQI-Metal Complexes, $(R_1, R_2\text{-DCNQI})_2M$; $(R_1, R_2 =$ $\text{CH}_3, \text{I}, M = \text{Ag}, \text{Cu}, \text{Li}_{1-x}\text{Cu}_x)$

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GROUND STATES OF DCNQI-METAL COMPLEXES, $(R_1, R_2\text{-DCNQI})_2M$; ($R_1, R_2 = \text{CH}_3, \text{I}$, $M = \text{Ag}, \text{Cu}, \text{Li}_{1-x}\text{Cu}_x$)

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Abstract The electrical resistance, magnetic susceptibility, ^1H - and ^{13}C -NMR measurements have been performed to investigate the ground states of the metal complexes of DCNQI, $(R_1, R_2\text{-DCNQI})_2M$, where $R_1(R_2)$ is CH_3 (Me) or iodine (I) and M is Ag, Cu or $\text{Li}_{1-x}\text{Cu}_x$. A new member of DCNQI-M complexes, $(\text{DI-DCNQI})_2\text{Ag}$, is nonmetallic below room temperature and shows a Bonner-Fisher type of susceptibility, of which the value is nearly the same as that of $(\text{DI-DCNQI})_2\text{Cu}$ at room temperature. An antiferromagnetic ordering was found at 5.5K in contrast to the spin-Peierls ground state in the isostructural $(\text{DMe-DCNQI})_2\text{Ag}$. The alloy systems, $(\text{DMe-DCNQI})_2\text{Li}_{1-x}\text{Cu}_x$, have been studied with an aim at control of the electronic state by the doping of Cu into the insulator, $(\text{DMe-DCNQI})_2\text{Li}$. For the Cu doping up to $x=30\%$, the systems show the spin-Peierls transition while the system with $x>50\%$ are metallic down to low temperatures. The $(T_1T)^{-1}$ of ^{13}C -NMR shows systematic change with doping.

INTRODUCTION

A family of $(2,5\text{-}R_1, R_2\text{-DCNQI})_2M$ contain a variety of electronic phases. Planer DCNQI molecules are uniformly stacked in one-dimensional (1D) columns along the crystallographic c -axis and, at the same time, coordinate to the metal ion, M , tetrahedrally through the cyano group [1]. The Li and Ag salts of DMe-DCNQI ($R_1=R_2=\text{Me}$; $M=\text{Li}$ or Ag), DMe-Li and DMe-Ag, behave metallic in high temperature region while, with decreasing temperature, the $4k_F$ charge density wave fluctuations associated with dimerization of DMe-DCNQI develop with appearance of nonmetallic resistive behavior and a spin-Peierls transition occurs at lower temperatures [2]. In this case, the metal ions do not contribute to the conduction band, which is constructed by the 1D π orbitals with quarter filling. On the other hand, in case of $M=\text{Cu}$ (DMe-Cu),

the 3d-orbitals of Cu ions are strongly hybridized with the π -band in the DMe-DCNQI columns [3, 4]. This salt remains metallic down to lower temperatures [5].

(DI-DCNQI)₂Cu ($R_1, R_2 = I$; $M = Cu$), DI-Cu, is also metallic down to low temperatures [6]. DI-Cu is compared with DMe-Cu in the following respects; 1) anisotropy of resistivity, $\rho_{\perp}/\rho_{\parallel} \approx 3$, is less than that of DMe-Cu ($\rho_{\perp}/\rho_{\parallel} \approx 10$) [7]; 2) the pressure-temperature phase diagram of DI-Cu is curious in that the critical pressure of the metal-insulator transition (~ 15 kbar) is much higher than the value of DMe-Cu (~ 0.3 kbar), and a metallic state appears again in a high pressure region (> 20 kbar) [7]; 3) spin susceptibility of DI-Cu is about twice larger than DMe-Cu and forms a broad peak around 110 K [8]. These characteristics of DI-Cu attract interest in the electronic state of the DI-DCNQI family.

In the present work, electrical resistance, R , magnetic susceptibility, χ , and 1H - and ^{13}C -NMR measurements have been performed to investigate the ground states of a series of $(R_1, R_2\text{-DCNQI})_2M$. We see the effect of the functional group, R_1 and R_2 , and filling and/or dimensionality of the electronic band possibly controlled by alloying of M ; the valence of Li and Ag in the salts is +1 while that of the Cu salt is $+1+\delta$, so that the Cu doping to the Li or Ag salt is expected to cause some change in filling of the 1D π -band and/or generate 3D character through hybridization of the Cu 3d orbitals with the π orbitals. In addition, we have synthesized and characterized a new salt, (DI-DCNQI)₂Ag, which is expected to be a purely π -electronic system analogous to DMe-Ag. In this paper, (i) the electronic state of DI-Ag is compared with DI-Cu and (ii) the doping effect on the insulating DMe-Li is examined.

The experiments in the present work were performed for powdered samples. The 1H - and ^{13}C -NMR measurements were made at frequencies of 85.3 and 61.5 MHz, respectively.

(DI-DCNQI)₂M [$M = Ag, Cu$]

The newly synthesized material, DI-Ag, is insulating below room temperature with a charge gap of ~ 490 K, which was obtained by the temperature dependence of resistance. The χ follows a Curie-Weiss law in high temperature region and forms a

broad peak around 30K. A sharp kink is observed around 10 K.

The ^1H -NMR shift, 1K , and the spectral width determined by the second moment scale to the static susceptibility above $\sim 50\text{K}$, the second moment continuous to increase; this may be a manifestation of the short range ordering. From the slope of the K - χ plot and linewidth- χ plot, it is found that the isotropic term and the square root of the mean square of the anisotropic term of the hyperfine coupling tensor, A , are $-230\text{Oe}/\mu_B$ and $270\text{Oe}/\mu_B$, respectively.

At 5.5K, the ^1H relaxation rate, $1/T_1^{-1}$, forms a peak anomaly and the ^1H -NMR linewidth abruptly becomes broadened as shown in Fig.1. These evidence an antiferromagnetic ordering of the spins at this temperature. The value of the linewidth extrapolated at 0 K is 260 kHz. The spectral width for a powdered sample with antiferromagnetic order is

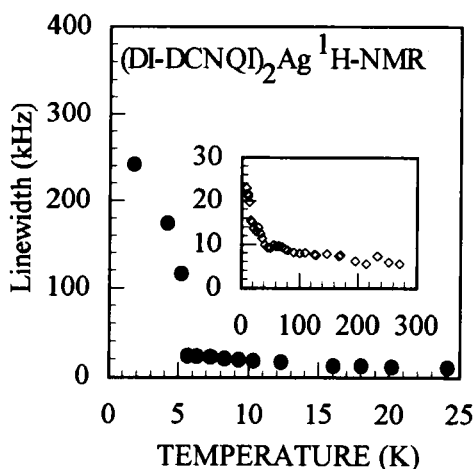


Fig 1 Temperature dependence of linewidth.

roughly given by $(\gamma A \langle \mu \rangle) / 2\pi$, where γ , A and $\langle \mu \rangle$ are the gyromagnetic ratio of ^1H nuclei, hyperfine coupling constant and amplitude of the AF moment, respectively. In our case, the external field is larger than the spin flop field so that the anisotropic part of A is an only contribution to the line width. The formula gives a small amplitude of $\langle \mu \rangle \sim 0.25\mu_B/\text{dimer}$, which may indicate reduction of the moment due to the zero point motion of spin waves, so-called spin contraction encountered in the low dimensional magnetic systems[9].

The DI-Ag is considered as a purely π -electron system like DMe-Ag and DMe-Li, of which the ground states are nonmagnetic spin-Peierls states associated with the 1D nature of the π -electron system. The DI-Ag salt is the first case in the DCNQI-M family that the π electrons are responsible for AF ground state. The difference of the

ground states are attributable to the difference in the dimensionality of the electronic states [10]; according to the estimation of the transfer integral [7, 11], the DI-systems are more three dimensional than the DMe-systems.

Next, our attention is directed to DI-Cu. In Fig.2, the spectral shift of ^{13}C - and ^1H -NMR, ^{13}K and ^1K from the line position of TMS are shown in comparison with χ [12]. Both of ^{13}K and ^1K form broad peaks like that of χ . However, it should be noted that the peak temperature of ^1K is lower than that of χ while the peak temperature of ^{13}K is slightly higher than that. The overall temperature dependence of χ is in between the two profiles. Since K

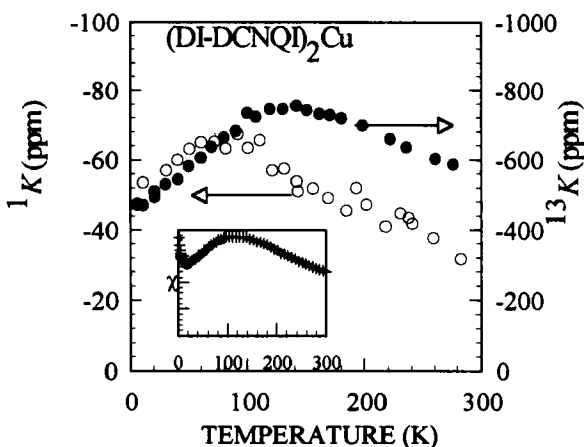


Fig.2 Temperature dependence of ^1H -shift, ^1K (open circles), and ^{13}C -shift, ^{13}K (closed circles). The shifts are measured from the line position of TMS. The χ is shown in the inset

probes the local spin susceptibility, this fact is considered as a microscopic evidence that the electronic structure consists of several bands with different characters, which is believed to come from hybridization of the π and d orbitals; ^1H -NMR detects the electrons of π -band preferentially, while ^{13}C -NMR at the cyano group coordinated to the Cu ions can probe the d-electrons through the off-site core polarization as well as the π electrons [13].

The ground state of DI-Ag salt is a magnetic insulator, while the isostructural DI-Cu salt is a paramagnetic metal with hybridization of the π orbitals with Cu 3d orbitals. The present result of DI-Ag shows an important role of π band in electron correlation manifesting itself in, e.g. enhancement of spin susceptibility in the DI systems. The first principles electronic band calculations show that the π band in the DI system is narrower than in the DMe system [14].



From the resistivity measurements of several alloy systems with different Cu contents, it was found that the systems up to $x \sim 30\%$ undergo the metal-insulator transition while the systems with $x > 50\%$ are metallic in the whole temperature range investigated. Figure 3 shows the temperature dependence of χ for typical alloy systems. A systematic change of behavior with the doping content is observed.

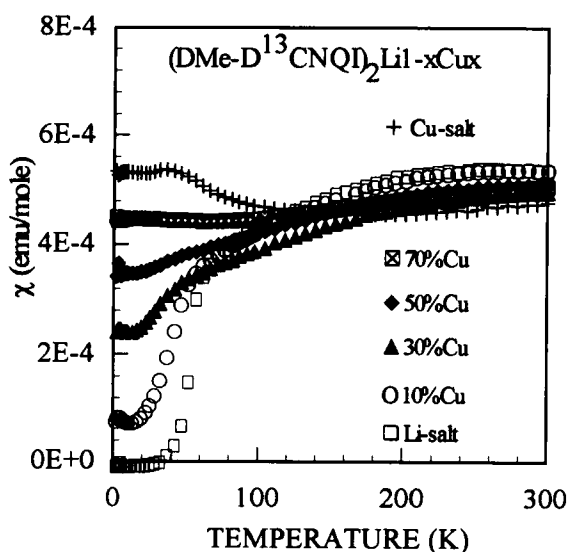


Fig 3 Temperature dependence of spin susceptibility.

The ^{13}C (in the cyano group) nuclear spin-lattice relaxation rate, $^{13}T_1^{-1}$, is shown in Fig.4. For $x < 30\%$, one can see abrupt decrease of $(^{13}T_1T)^{-1}$, which is associated with the spin-Peierls transition. The transition temperature slightly shifts with increasing x .

For the Cu-rich systems of $x > 50\%$, $(^{13}T_1T)^{-1}$ does not exhibit any anomaly but converges into about $0.03 \text{ sec}^{-1}\text{K}^{-1}$ in the low-temperature limit. They show positive temperature dependence, which have a clear correlation with the doping content. This is an indication that the doping causes some change in the electronic states, particularly in

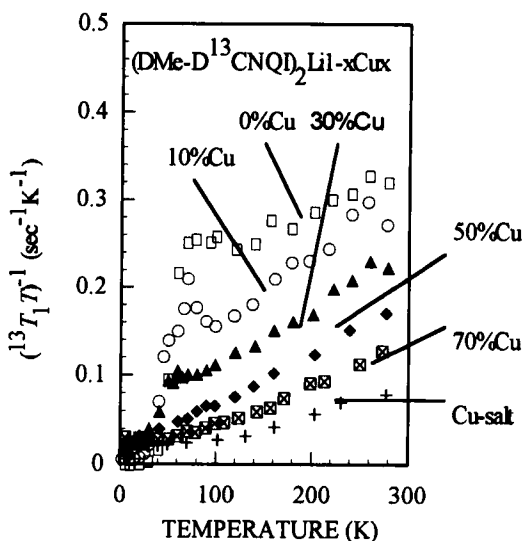


Fig 4 Temperature dependence of ^{13}C -NMR, $(^{13}T_1T)^{-1}$.

excitation spectrum visualized in higher temperature region.

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